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CHAPTER 12 STOICHIOMETRY REVIEW
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Important Vocabulary • Stoichiometry-
The calculations of quantities in chemical
reactions in a subject of chemistry • Mole
ratio- a conversion factor derived from the
coefficients of a balanced chemical equation
interpreted in terms of moles • Limiting
reagent- The reactant that determines the
amount of product that can be formed by a
reaction • Excess ..., Stoichiometry
CHAPTER 12 What You™ll Learn You will
write mole ratios from balanced chemical
equations. You will calculate the number of
moles and the mass of a reactant or product
when given the number of moles or the mass
of another reactant or product. You will
identify the limiting reactant in a chemical
reaction. You will determine the percent yield
of a chemical reaction. Why It™s ...,
Stoichiometry 379 CHAPTER 12
Assessment 36. a. Two formula units KClO_3
decom- pose to form two formula units KCl
and three molecules O_2 . b. Four molecules

NH_3 react with six molecules NO to form
i-ve mol-, Chapter 12 Stoichiometry Review
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