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Personal beliefs influence health behavior.,

14.2 The Gas Laws > 12 Copyright ©

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A decrease in pressure at constant

temperature must correspond to a

proportional increase in volume. € The

calculated result agrees with both kinetic

theory and the pressure-volume relationship.,

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Quizlet., Chapter 14: Self-Defeating

Thoughts and Behaviors 137 choices that

this belief creates have on you, the people in

your life, and your Recovery?

Pathfinder™s Exercise I 1) Pick two

self-defeating behaviors that you most

closely identify with. 2) Give an example of

how using one behavior harms you, the

people in your life, and your Recovery., List

four everyday items that rely on the behavior

of gases to operate properly. SECTION 12.2

FACTORS AFFECTING GAS PRESSURE

(pages 330€332) This section explains

how gas pressure is affected by the amount

of gas, the volume of a container, and

temperature changes. Amount of Gas (pages

330€331) 1., _____ 14. The volume of a

gas is doubled while the temperature is held constant. The pressure of the gas a. remains unchanged. c. is doubled. b. is reduced by one half. d. depends on the kind of gas.

_____ 15. As the temperature of the gas in a balloon decreases a. the volume increases.
b. the pressure increases.

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