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SECTION 22-1 OBJECTIVES Explain what

a nuclide is, and describe the different ways

nuclides can be represented. Define and

relate the terms ... 702 CHAPTER 22

FIGURE 22-1 This graph shows the

relationship between binding energy per

nucleon and mass num-, 1 1 Chapter 21

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ixvlrq, Chapter 23: Nuclear Chemistry 23.1:

The Nature of Nuclear Reactions Nucleons: -

the particles that make up a nucleus of an

atom (protons,(1 1 p + or 1 1 H) and

neutrons, ($1\ 0\ n$)). Isotopes: - atoms that have different mass number but the same atomic number or number of protons. ... $^{22}_{20}\text{Ti}$ + $0\ ^{-1}\text{e}$ + 1, Chapter 22 notes chapter 22 nuclear chemistry section 22 , view notes chapter 22 notes from chemistry honors che at madison high, madison, nj chapter 22: nuclear chemistry section 22 1: the nucleus atomic nuclei= protons and neutrons (together, \propto Prob: An atom has 20 protons and 22 neutrons. Draw the ... \propto Nuclear reactions are not affected by any chemical changes (temp, pressure, catalysts). \propto Nuclear reactions cannot be sped up, slowed ... Chapter 25 \propto Nuclear Chemistry Author: Authorized User Created Date:, Chapter 21 Nuclear Chemistry Chapter 21 \propto Assignment A: Natural Radioactivity: Where Does It Come From? ... of the chapter. Questions, Exercises, and Problems 30 \propto 32. Include Questions 33 \propto 35 if assigned by your instructor. Check your answers with those at the end of the

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