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that produce an increase in $[H^+]$ when
dissolved in water bases - compounds that
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The correct way of making a solution of acid
in water is to (a) add water to acid. (b) add
acid to water. (c) mix acid and water
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this resonance occurs at rather small
chemical shift€”typically around δ 1. In
aromatic amines, this resonance is at greater
chemical shift, as in the second of the
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system is essentially no different than the,

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axis of the DNA molecule, minimizing their
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class of Arrhenius acids which are capable of
donating two protons or hydrogen cations per

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