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Stoichiometry EXAMPLE Problem 11.1  
Interpreting Chemical Equations The  
combustion of propane (  $C_3H_8$  ) provides  
energy for heating homes, cooking food, and  
soldering metal parts. Interpret the equation  
for the combustion of propane in terms of  
representative particles, moles, and mass., 1  
AP CHEMISTRY CHAPTER REVIEW  
CHAPTER 11: RATE OF REACTION You  
should understand the definition of reaction  
rate, as well as how rates might be  
measured in a laboratory setting. You should  
know the difference between average rate,  
instantaneous rate, and initial rate., Section  
Review, p. 367  $P_1V_1 = P_2V_2$   $P_2 = \dots$   
MODERN CHEMISTRY CHAPTER 11  
SOLUTIONS MANUAL 2 ..., 1.11 Dipole  
Moments and Molecular Polarity 1.12

Intermolecular Forces and Physical  
Properties 1.13 Solubility ... 2 CHAPTER 1 A  
Review of General Chemistry, 11. An atom is  
said to be in an excited state when it  
possesses more than its minimum energy (its  
ground state). An atom is promoted from its  
ground state to an excited state by the  
absorption of energy, and returns to its  
ground state by the emission of the excess  
energy. 12., Chemistry 11 Answer Keys;  
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duration (e.g. 50 minutes), no open book or  
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table included with the exam and the  
calculator that you will be using for the  
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