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temperature T , the average thermal energy
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 2.07×10^{-21} J, or 0.013 eV, at room
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equilibrium is the state in which both
reactants and products are present in
concentrations which have no further
tendency to change with time, so that there is
no observable change in the properties of the
system. Usually, this state results when the
forward reaction proceeds at the same rate
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